

Testing a Hypothesis (Significance Tests)

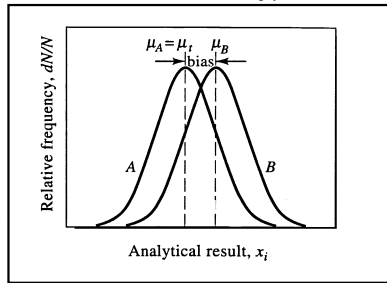
Carry out measurements on an accurately known standard.

Experimental value is different from the true value.

Is the difference due to a systematic error (bias) in the method - or simply to random error?

Assume that there is *no* bias (NULL HYPOTHESIS), and calculate the probability that the experimental error is due to random errors.

Figure shows (A) the curve for the true value ($\mu_A = \mu_t$) and (B) the experimental curve (μ_B)



$$\text{Bias} = \mu_B - \mu_A = \mu_B - x_t$$

Test for bias by comparing $\bar{x} - x_t$, with the difference caused by random error

Remember confidence limit for μ (assumed to be x_t , i.e. *assume no bias*) is given by:

$$\text{CL for } \mu = \bar{x} \pm \frac{ts}{\sqrt{N}}$$

\therefore at desired confidence level, random errors can lead to:

$$\bar{x} - x_t = \pm \frac{ts}{\sqrt{N}}$$

\therefore if $|\bar{x} - x_t| > \left| \frac{ts}{\sqrt{N}} \right|$, then at the desired confidence level bias (systematic error) is likely (and vice versa).

Detection of Systematic Error (Bias)

A standard material known to contain 38.9% Hg was analysed by atomic absorption spectroscopy. The results were 38.9%, 37.4% and 37.1%. At the 95% confidence level, is there any evidence for a systematic error in the method?

$$\begin{aligned} \bar{x} &= 37.8\% & \therefore \bar{x} - x_t &= -1.1\% \\ \sum x_i &= 113.4 & \sum x_i^2 &= 4208.30 \\ \therefore s &= \sqrt{\frac{4208.30 - (113.4)^2/3}{2}} = 0.943\% \end{aligned}$$

Assume null hypothesis (no bias). Only reject this if

$$t_{\text{calc}} = (\bar{x} - \mu) \frac{\sqrt{N}}{s} \quad \text{Reject if } t_{\text{calc}} > t_{\text{table}}$$

But t (from Table) = 4.30, s (calc. above) = 0.943% and $N = 3$

$$2.02 < 4.30$$

Therefore the null hypothesis is maintained, and there is no evidence for systematic error at the 95% confidence level.

Are two sets of measurements significantly different?

Suppose two samples are analysed under identical conditions.

Sample 1 $\rightarrow \bar{x}_1$ from N_1 replicate analyses

Sample 2 $\rightarrow \bar{x}_2$ from N_2 replicate analyses

Are these significantly different?

Using definition of pooled standard deviation, the equation on the last overhead can be re-arranged:

$$t_{\text{calc}} = \frac{\bar{x}_1 - \bar{x}_2}{s_{\text{pooled}} \sqrt{\frac{N_1 + N_2}{N_1 N_2}}}$$

Again reject if $t_{\text{calc}} > t_{\text{table}}$

Detection of Gross Errors

A set of results may contain an outlying result - out of line with the others.
Should it be retained or rejected?
There is no universal criterion for deciding this.
One rule that can give guidance is the *Q test*.

The parameter Q_{calc} is defined as follows:

$$Q_{calc} = \frac{|outlier - nearest_neighbor|}{total_spread}$$

Again reject if $Q_{calc} > Q_{table}$

No. of observations	90%	95%	99%	confidencelevel
3	0.941	0.970	0.994	
4	0.765	0.829	0.926	
5	0.642	0.710	0.821	
6	0.560	0.625	0.740	
7	0.507	0.568	0.680	
8	0.468	0.526	0.634	
9	0.437	0.493	0.598	
10	0.412	0.466	0.568	

Rejection of outlier recommended if $Q_{calc} > Q_{table}$ for the desired confidence level.

- Note:**
1. The higher the confidence level, the less likely is rejection to be recommended.
 2. Rejection of outliers can have a marked effect on mean and standard deviation, esp. when there are only a few data points. *Always try to obtain more data.*

Q Test for Rejection of Outliers

The following values were obtained for the concentration of nitrite ions in a sample of river water: 0.403, 0.410, 0.401, 0.380 mg/l. Should the last reading be rejected?

$$Q_{calc} = \frac{|0.380 - 0.401|}{(0.410 - 0.380)} = 0.7$$

But $Q_{table} = 0.829$ (at 95% level) for 4 values

Therefore, $Q_{calc} < Q_{table}$ and we cannot reject the suspect value.

Suppose 3 further measurements taken, giving total values of:

0.403, 0.410, 0.401, 0.380, 0.400, 0.413, 0.411 mg/l. Should

0.380 still be retained?

$$Q_{calc} = \frac{|0.380 - 0.400|}{(0.413 - 0.380)} = 0.606$$

But $Q_{table} = 0.568$ (at 95% level) for 7 values

Therefore, $Q_{calc} > Q_{table}$ and rejection of 0.380 is recommended.

But note that 5 times in 100 it will be wrong to reject this suspect value!

Also note that if 0.380 is retained, $s = 0.011$ mg/l, but if it is rejected, $s = 0.0056$ mg/l, i.e. precision appears to be twice as good, just by rejecting one value.